ORGANIC CHEMISTRY 2 LECTURE GUIDE 2019

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Lesson IV.12. Substituent Effects on the Regiochemistry of EAS

More stable intermediate = more product

If there is already a substituent on a benzene ring and we then attempt to further substitute it, we need to consider what is the major product will be. If a monosubstituted arene is subjected to EAS, we can get o-, m-, and p- products:

monosubstituted arene is subjected to EAS, we can get o -, m -, and p - p	
A	
In order to understand which product(s) are formed in highest yiel examine the mechanism, keeping in mind one of the <i>generally principles</i> used in examining organic chemical transformations:	
B	
*EAS = Electrophilic aromatic substitution	
.T L	
<u>Notes</u>	

Consider a resonance donor (Z) substituent with electrophile (E) added o-, m- or p-:

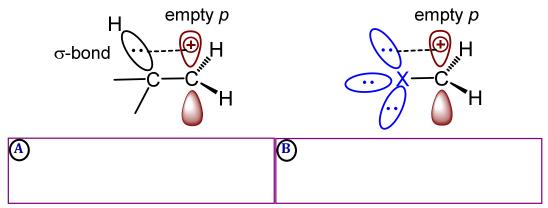
<u>Notes</u>			

<u>Notes</u>		

Next, a halogen (X = F, Cl, Br, I):

<u>Notes</u>			

For halogens consider their interaction with an adjacent carbocation as compared to the hyperconjugation that stabilizes carbocations:



The figure above illustrates that a halogen is able to stabilize a positive charge on an adjacent C. With this knowledge and the resonance contributors on the previous page, we conclude:



<u>Notes</u>			

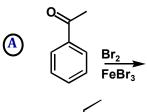
Finally, consider an inductive withdrawing group (Z):

<u>Notes</u>		

	O			0					correlate	a	substituent's	EAS
Re	eactivity	Group) with it	s EAS D	ire	ecting	g T'yr	oe:				

Reactivity Grou	up with its EAS Directing Type:
A	
and	
B	
	t these are <i>general trends</i> to help us predict the major product; ll not get <i>exclusively</i> the favored product(s)
<u>Notes</u>	
VULES	

Example: Give the major product(s) of the following reactions using your knowledge of whether the existing substituent is o-/p- directing or m-directing.



$$\begin{array}{c|c}
\hline
B & \hline
 & HNO_3 \\
\hline
 & H_2SO_4
\end{array}$$

<u>Notes</u>		